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Synthesis of Phosphinines and its Saturated Devivatives:DI-, Tetra-and Hexahydrophosphinine Oxides

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SYNTHESIS OF PHOSPHININES AND ITS SATURATED DERIVA-TIVES: DI-, TETRA- AND HEXAHYDROPHOSPHININE OXIDES

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Our ring enlargement procedure 1 is extended to the synthesis of valuable 1,2-dihydrophosphinine oxides ($\underline{3}$) and 1,2,3,6-tetrahydrophosphinine oxides ($\underline{4}$) with alkoxy substituent on the P-atom. Dichlorocarbene is added to the double bond of dihydro-1H-phosphole oxide $\underline{1}$ in the first step. The diastereoisomeric adducts ($\underline{2}$ and $\underline{2}^{\times}$) so obtained afford the ring expanded products ($\underline{3}^2$ or $\underline{4}$) by the opening of the cyclopropane ring (Scheme I).

$$R^{10} \longrightarrow R^{2} : CCl_{2}$$

$$R^{10} \longrightarrow R^{2} : CC$$

Scheme I

Depending on the structure of the starting material (R^2 may be Me or H in $\underline{1}$), $\underline{3}$ and $\underline{4}$ may be formed as a single product

 $(R^3 = R^4 = Me)$, or as a mixture of two double bond isomers $(R^3 = H, R^4 = Me)$ and $R^3 = Me$, $R^4 = H$.

Tetrahydrophosphinine oxide $\underline{4}$ may exist under the equilibrium of boat or half-chair conformers. The predominance of a boat conformer was substantiated for the 3,5-dimethyl-product, while the involvement of a half-chair conformer was pointed out in the case of the 3- and 5-methyl-derivatives.

The dihydrophosphinine oxide (3) gives adduct 5 on reaction with a second unit of dichlorocarbene to result 4-dichloromethylene-1,4-dihydrophosphinine oxide 6 or phosphepine oxide 7 by the opening of the cyclopropane ring (Scheme II).

Cl Cl
$$R^3$$
 from $R^3 = R^4 = Me$
 $R^3 = R^4 = Me$

Scheme II

The outcome of the reaction depends on the number of the skeletal methyl groups in intermediate $\underline{5}$. $\underline{7}$ is formed by ring expansion (only one of R^3 and R^4 is Me), while $\underline{6}$ by the rather unusual opening of the cyclopropane ring (both R^3 and R^4 are Me).

The 1,2-dihydrophsphinine oxides ($\underline{3}$) are also useful in the synthesis of other phosphinine derivatives. E.g. hexahydrophosphinine oxide $\underline{8}$ is formed from $\underline{3}$ on reaction with three equivalents of hydrogen. The resulting phosphinic acid ester ($\underline{8}$, R 3 =H, R 4 =Me) can then be transformed to the amide ($\underline{9}$) (Scheme III).

Scheme III

The 3-methyl-product is formed as the mixture of two diastereoisomers, while the 3,5-dimethyl-derivative as three isomers. Although the trans and cis diastereoisomers of $\underline{8}$ (R³=H, R⁴=Me) and $\underline{9}$ can be imagined as the equilibria of two pairs of conformers, the measured values for the 3 J(P-C) coupling constants 3 show the equilibria to be strongly biased toward the form with equatorial skeletal methyl group (Scheme IV).

H₃C

$$Z = P$$
 $Z = P$
 $Z =$

$$Y = OR^{1}(8, R^{3} = H, R^{4} = Me), NMe, (9)$$

Scheme IV

Fully unsaturated (aromatic) species, phosphinines ($\underline{12}$) can also be obtained from the P-alkoxy-dihydrophosphinine oxides ($\underline{3}$). Phosphinic chloride $\underline{10}$ is prepared in the first step to give the expected phosphinine ($\underline{12}$, R⁵=H or Me) after reduction and dehydrochlorination (Scheme V).

The formation of the dimethyl-phosphinine is accompanied by side reactions (decomposition or overreduction to 3,5-dimethylphosphinine).

$$R^{3} \xrightarrow{PCl_{5}} R^{4} \xrightarrow{PCl_{5}} R^{3} \xrightarrow{Cl} R^{4} \xrightarrow{PhSiH_{3}} R^{3} \xrightarrow{Cl} R^{4} \xrightarrow{PhSiH_{3}} R^{3} \xrightarrow{Cl} R^{4} \xrightarrow{PhSiH_{3}} R^{5} = H, Me$$

$$R^{3} \xrightarrow{R^{4}} R^{4} \xrightarrow{R^{5}} R^{4} \xrightarrow{R^{5}} R^{5} = H, Me$$

$$R^{5} = H, Me$$

$$R^{5} = H, Me$$

$$R^{5} = H, Me$$

$$R^{5} = H, Me$$

Structure of the products ($\underline{2}$ - $\underline{4}$, $\underline{6}$ - $\underline{9}$ and $\underline{12}$) was confirmed by 1 H, 13 C and 31 P NMR and mass spectroscopy.

The 1,2- and 1,4-dihydrophosphinine oxides ($\underline{3}$ and $\underline{6}$) display interesting biological activity.

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